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# **RESEARCH REPORT**

6666 West Quincy Avenue Denver, Colorado 80235

**SUBJECT AREA: Water Treatment and Operations** 



# TABLE OF CONTENTS

			Page
List of Tables			
List of Figures			V
Foreword			vii
Abstract			ix
Executive Summary		• · · · · ·	X
Introduction			xi
Literature review			1
Origins of Disinfection Effort with	Chloneri		3
Chloramine Chemistry	chioramines		3
Disinfection by Chloramines		ана страната на страната н Страната на страната на стр	3
Chemical Reactivity of Chloraminos			8
Biological Reactions Leading to Mior	abial T		11
Materials	obial Inactivati	on	13
Bacteria			19
Culture Media, Buffers, Chemicals and	d Untra		19
Demand-Free Glassware	u water		19
Methods	· ·		19
Disinfectant Preparation, Measurement			21
Microbiological Methods	and Stability		21
Chemical Reactivity with Biochemical	Carl - Land		26
Results	Substrates		35
Monochloramine Stability			39
Inactivation Studies			39
Nucleic Acid Studies	_		44
Action on the Cellular Envelope			49
Protein-Mediated Systems and Amino An	4.1 D		62
Discussion Joseph and Amino Ac.	id Reactions		69
Disinfectant Stability			87
Monochloramine Reactions with Biochem	inal ( 1 .		87
Action on the Cell Envelope	ical Substrates		87
Reactions with Proteins and Protein-Me	diated D		93
Multi-Hit Model	ediated Processes	<b>3</b>	94
Applied Aspects			96
Conclusions			97
iterature Cited			99
Appendix A			101
Appendix B			117
oppendix C	)		119
			121

#### LIST OF TABLES

### Table Page 5 1. Physical and chemical properties of inorganic chloramines. 9 2. Dosages of various chlorine species required for 99 percent inactivation of Escherichia coli and poliovirus I. 3. Calculated molar absorption coefficients for 25 monochloramine and dichloramine at 260 nm and 300 nm. 4. Times for E. coli incubation with radiolabeled 34 substrates as used in standard curve generations and uptake inhibition experiments. 5. Half-life and first-order rate constants for the decay of 41 monochloramine as a function of temperature and pH. 45 6. Stability of monochloramine under volatility-limiting conditions and volatility-promoting conditions. 7. Percent distribution of E. coli by clump size. 46 8 Percent increase in E. coli densities after various 47 chemical and physical treatments. 9. 57 Consumption of monochloramine by nucleic acids and their components. 65 10. Change in nucleotide concentration due to the reaction with monochloramine and free chlorine (analysis by HPLC). 75 11. Consumption of monochloramine by amino acids and nucleic acids. 77 12. Oxidation of cysteine residues by monochloramine. 78 13. Oxidation of cysteine to cystine by monochloramine. 83 14. Oxidation of sulfhydryl groups of both intact bacteria and bacterial extracts of E. coli. 86 15. Changes in tryptophan residue absorbance in soluble proteins of E. coli after exposure to monochloramine. 89 16. Chemical reactivities of monochloramine and free chlorine

v

with various biological materials.

## LIST OF FIGURES

Page

- Ar 11, mpt

111 F. 1

-

## Figure

ere;

1.	Standard curve for monochloramine concentrations using syringaldazine.	23
2.	Uptake of D-(U-Y\C)glucose by <u>E. coli</u> .	33
3.	Stability of monochloramine at pH 10 at various temperatures.	40
4.	Stability of monochloramine at various pH.	42
5.	Stability of monochloramine at pH 10 when stored in the dark and in light.	43
6.	Inactivation of <u>E. coli</u> B by 1, 2.5, 4, and $10 \text{mg/l}$ of monochloramine.	48
7.	Effect of monochloramine on the transforming activity of DNA from Haemophilus influenzae.	50
8.	Effect of long contact time of monochloramine on the transforming activity of DNA from <u>Haemophilus</u> influenzae.	51
9_	Nucleic acid and its structural components.	53
10.	. Comparative consumption of monochloramine by yeast RNA and yeast DNA.	54
11.	Comparative consumption of monochloramine by adenine and cytosine nucleotides, nucleosides and free bases.	55
12.	Comparative disappearance of the 257 nm absorption band of yeast RNA after exposure to monochloramine and free chlorine.	58
13.	Comparative disappearance of the maximum absorption bands of RNA nucleotides after exposure to monochloramine and free chlorine.	59
14.	Comparative disappearance of the maximum absorption bands of RNA purine and pyrimidine bases after exposure to monochloramine and free chlorine.	60
15.	Chromatograms of AMP after treatment with monochloramine and free chlorine.	61
16.	Comparative changes in absorbance at 525 nm after treatment of <u>E. coli</u> with monochloramine and free chlorine.	63
17.	Leakage of nucleic acids and other UV-absorbing material	66

## LIST OF FIGURES (cont'd)

## Page Figure 67 18. Leakage of []S-proteins and smaller []S-molecules. 19. Inhibition of uptake of various radiolabeled substrates. 68 20. Inhibition of oxygen consumption (glucose and succinate 70 as substrates). 21. Inhibition of oxygen consumption (pyruvate and acetate as 71 substrates). 72 22. Inhibition of total dehydrogenase activity. 74 23. Inhibition of succinic dehydrogenase activity. 24. Repetitive ultraviolet scans of the reaction of monochloramine 80 with cysteine at a molar ratio of 2 to 1, CYSH to NH\*Cl. 25. Repetitive ultraviolet scans of the reaction of monochloramine 81 with cysteine at a molar ratio of 0.75 to 1, CYSH to NH\*Cl. 26. Percent -SH groups recovered after the reaction of monochloramine 82 with cysteine at various molar ratios. 85 27. Oxidation of bacterial thiol groups.

## FOREWORD

This report is part of the on-going research program of the AWWA Research Foundation. The research described in the following pages was funded by the Foundation in behalf of its members and subscribers in particular and the water supply industry in general. Selected for funding by AWWARF's Board of Trustees, the project was identified as a practical, priority need of the industry. It is hoped that this publication will receive wide and serious attention and that its findings, conclusions, and recommendations will be applied in communities throughout the United States and Canada.

The Research Foundation was created by the water supply industry as its center for cooperative research and development. The Foundation itself does not conduct research; it functions as a planning and management agency, awarding contracts to other institutions, such as water utilities, universities, engineering firms, and other organizations. The scientific and technical expertise of the staff is further enhanced by industry volunteers who serve on Project Advisory Committees and on other standing committees and councils. An extensive planning process involves many hundreds of water professionals in the important task of keeping the Foundation's program responsive to the practical, operational needs of local utilities and to the general research and development needs of a progressive industry.

All aspects of water supply are served by AWWARF's research agenda: resources, treatment and operations, distribution and storage, water quality and analysis, economics and management. The ultimate purpose of this effort is to assist local water suppliers to provide the highest possible quality of water, economically and reliably. The Foundation's Trustees are pleased to offer this publication as contribution toward that end.

The purpose of this project was directed at determining the mechanisms of inactivation of combined chlorine on bacteria. With many utilities using combined chlorine rather then free chlorine, it is important to know and understand the mechanisms of bacterial disinfection so that treatment practices can be improved. Chloramines and chlorine were found to react differently with the nucleic components of bacteria and that some chloramine reactions with cellular components may be reversible.

Jerome B. Gilbert Chairman, Board of Trustees AWWA Research Foundation

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ix

## ABSTRACT

A study was undertaken to provide a more perspicuous understanding of the mechanism of bacterial inactivation by monochloramine. The chemical reactivity of the disinfectant with various biochemical substrates was studied at 25°C and pH 7.0. The disinfectant readily reacts with 4 compounds: cysteine, cystine, methionine and tryptophan. Rapid reactions with only these compounds provided a partial explanation for its lesser biocidal efficiency when compared to free chlorine, which reacts with a wide variety of substrates. The extremely slow chemical and biological reactions with nucleic acids ruled out these bacterial components as primary lethal targets. Monochloramine did not severely damage the cell envelope of Escherichia coli B as indicated through cell lysis and cell leakage experiments. Since monochloramine-reactive compounds were amino acids, it appeared that the mode of action of the disinfectant involved proteins or protein-mediated reactions. Inhibition of typical protein-associated biological activities such as bacterial transport, respiration and substrate dehydrogenation were observed at concentrations normally employed under disinfection conditions. However, in all cases, the kinetics of bacterial inactivation were not consistent with the kinetics of biological inhibition. The inactivation of E. coli B was generally more rapid than the inhibition of biological activity.

Inactivation curves for <u>Escherichia coli</u> B were characterized by initial shoulders followed by exponential kill. Bacterial clumping and time required for the disinfectant to penetrate the cell and promote leakage were ruled out as possible causes for the initial lag in the curves. The observed differences in rates of kill and inhibition of various biological activities suggested that the shoulder formation was due to reactions at many sensitive sites in the bacterium which preceded inactivation. The mode of action of monochloramine appeared to involve multiple hits by the disinfectant on the bacterial cell before death occurred.

x